

1-есеп.

$\text{NH}_3 + \text{H}_2\text{O} = \text{HNO}_3 = 20\%$	100% — 80г
$m = 80г$	20% — x
$m(\text{H}_2\text{O}) - ?$	$\frac{80 \cdot 20}{100} = 16г$
$m(\text{HNO}_3) - ?$	$m(\text{H}_2\text{O}) = 1.2 + 16 = 18г$

2-есеп.

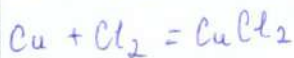
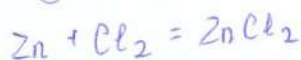
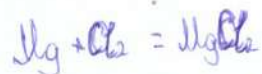
$\text{C}_2\text{H}_4\text{O}_2 = 32$

$M = 12 \cdot 2 + 1 \cdot 4 + 16 \cdot 2 = 60$

$V = \frac{m}{M} = \frac{32}{60} = 0,05 \text{ моль}$

$m(\text{C}_2\text{H}_4\text{O}_2) = 32$	ш:
$v(\text{C}_2\text{H}_4\text{O}_2) = 0,05 \text{ моль}$	
$V - ?$	

3-есеп.





Main content area for solutions, currently blank.