

1 тапсырма

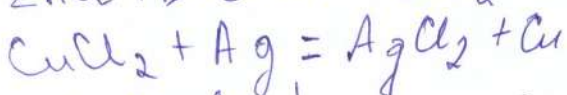
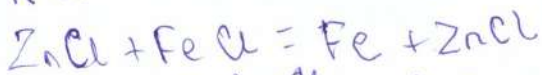
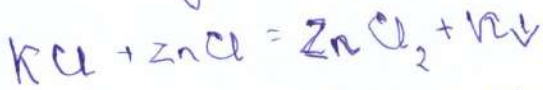
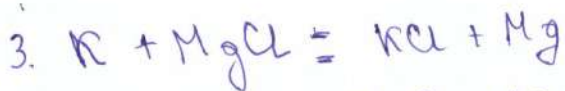
$$m(\text{H}_2\text{O}) = 29 \text{ г/моль} \quad 32 \text{ электрон}$$

$$m(\text{аммиак}) = 80 \text{ г}$$

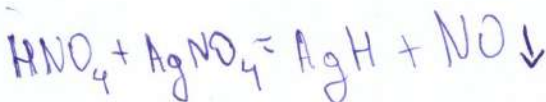
2 тапсырма

$$1. M_r(\text{C}_2\text{H}_4\text{O}_2) = 12 \cdot 2 + 1 \cdot 4 + 16 \cdot 2 = 60$$

$$Q_{\text{mol}} = \frac{m(\text{C}_2\text{H}_4\text{O}_2)}{M_r(\text{C}_2\text{H}_4\text{O}_2)}$$



$$4. \frac{2 \text{ Ar}(\text{Me})}{2 \text{ Ar}(\text{Me}) \cdot 3,82 \cdot 10^{-24}} = 4 \text{ Ar}(\text{Me}) \cdot 3,82 \cdot 10^{-24} = 1,33$$



$$S = \frac{40,50690}{1} = 1,33$$

$$O = 30,38\%$$

$$C = \text{H}$$

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Қатысушының шешімдерін толтыруға арналған өріс / Поле для заполнения решений участника

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